AGASYAN, E.O.

Pulse stretcher with logarithmic characteristics. Prib. 1 tekn. eksp. 8 no.6:103-105 N-D '63. (MIRA 17:6)

1. Fizicheskiy institut AN ArmSSR.

ACCESSION NR: AP4010027

8/0022/63/016/006/0125/0130

AUTHORS: Agasyan, B. O.; Marikyan, G. A.

TITLE: Investigations of ionization fluctuation in proportional low pressure counters

SOURCE: AN ArmSSR. Izvestiya. Ser. fiz.-matem. nauk, v. 16, no. 6, 1963, 125-130

TOPIC TAGS: ionization fluctuation, prossure counter, mu meson, cosmic ray, Geiger counter, propane gas, proportional counter

ABSTRACT: The ionization fluctuation curve width has been studied as a function of mixture parameters in a proportional counter for high-energy μ - mesons of cosmic origin. The apparatus consists of a Geiger counter telescope, two proportional counters, and electronic instruments to measure the pulse amplitude and μ -meson trajectory through the telescope. The counter is filled with a mixture of 25% argon and 75% methane at μ 0 cm Hg pressure. The measured fluctuation width was found to be μ 0% (μ 2.5%) compared to a 31% estimate from the theory of L. Landau (On the Energy Loss of Fast Particles by Ionization. Jour. Phys. (USSR) 8, 201,

Card 1/2

ACCESSION NR: AP4010027

1944). Measurements were also obtained in pure propane at 1.0, 0.5, and 0.25 atm pressure with 42%, 49% and 65% fluctuation widths respectively. It is shown that the fluctuation width decreases as the gas pressure in the counter increases. Further measurements at higher pressures are deemed necessary. "The authors are thankful to V. M. Kharitonov for his interest in this work." Orig. art. has: 6 figures.

ASSOCIATION: none

SUBMITTED: 11Jan63

DATE ACQ: 03Feb64

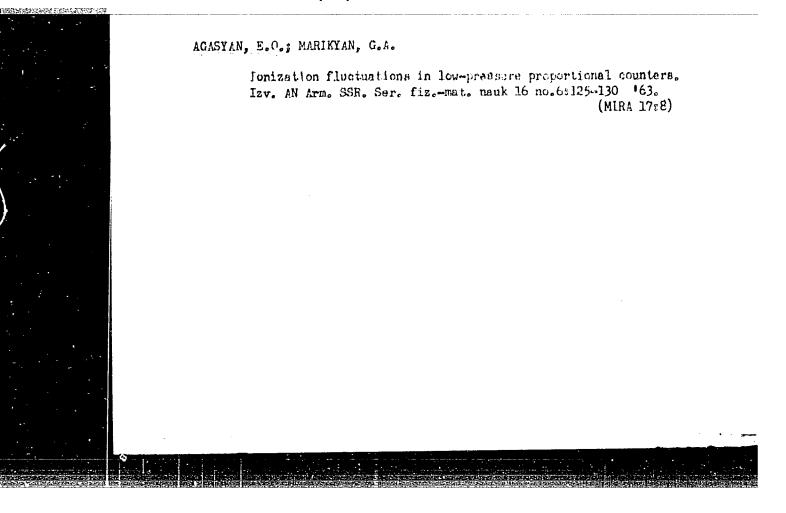
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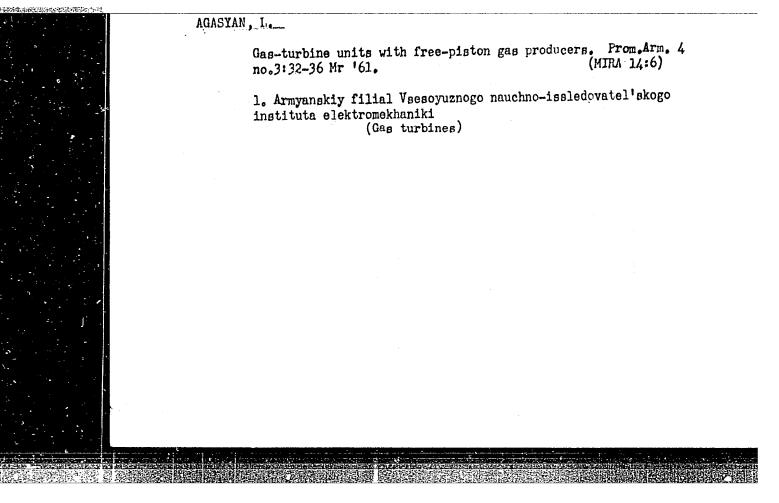
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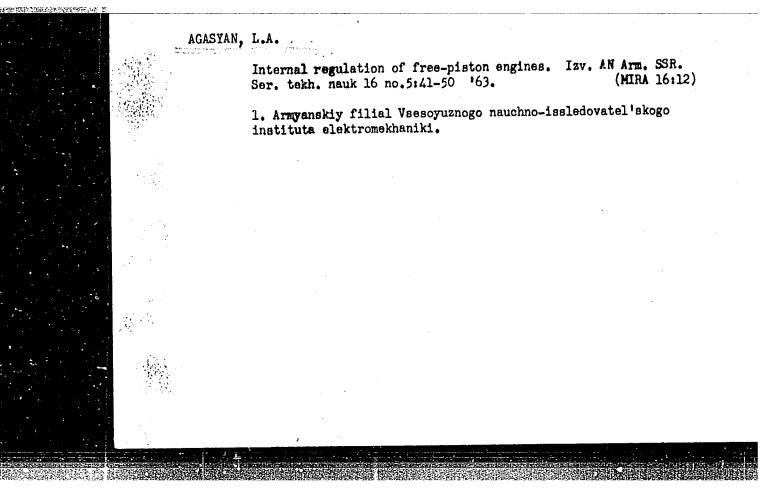
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ACCESSION MR: APSO18707

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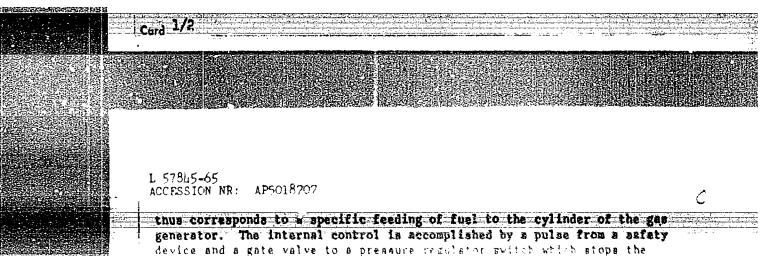
AUTHOR: Agasyan, 1. //

Time Pegulation of a was-turbine electron weem electrones to a thia free-platon.

SOURCE: AN Arm SSR. Izvestiya. Seriya tekhnicheskikh nauk. v. 17, no. 5, 1964, 43-50

TOPIC TAGS: automatic control, automatic control system, electric power engineering, gas turbine engine, thermoelectric generator, gasoline engine

ABSTRACT: The task of the control system is to assure stability of operation of a free-piston gas generator with a tortus of A when the fraction of the load of the installation. The present article proposes a simple method



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	Orig. art. hast 3 figures, 11 f	ormulas.	
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\$/075/62/017/005/004/007 1033/1233

AUTHORS:

Golovina, A.P., Alimarin, I.P., Bozhevol'nov, Ye. A.

and Agasyan, L.B.

TITLE:

Datiscine - a newfluorimetric reagent for zirconium

PERIODICAL:

Zhurnal analiticheskoy khimii, v.17, no. 5, 1962,

591-594

Datiscine (3,5,7,2' - tetraoxyflavone glucoside) produces with a number of cations soluble compounds which fluocresce upon irradiation with ultraviolet rays. In the case of Zr maximal fluorescence is observed in a 6 N HCl medium at 520 mm upon irradiation at 388 mp. A 100 fold excess of reagent is permissible. Intensity of fluorescence reaches a constant value

card 1/3

\$/075/62/017/005/004/007 IO33/I233

Datiscine - a new

after 15-20 min. It decreases with increase of ethanol concentration. Intensity is proportional to Zr concentration in the range of 0.005 - 3 ml. This makes datiscine a most sensitive reagent for Zr. In the 1.5 - 6 N HCl acidity range there is no interference from any amount of Mg and Zn, 100,000-told excess of Al and 100-fold excesses of Ag, Cd, Mn(II), Cu(II), Pb, Hg(II), Be, Co(II), In, Cr(III), Fe(II), Ta(V), V(V), Ni(II), Nb(V), W(VI), Y, U(VI), Ce(III), and La. In 6 N HCl, 10-fold excesses of Fe(III), Mo(VI), Ti(VI), Sb(V), Th and Ga do not interefere. This method was used for determination of Zr in aluminum and magnesium alloys. There are 2 figures and 4 tables.

Card 2/3

s/075/62/017/005/004/007 1033/1233

Datiscine - a new.

ASSOCIATION: Moskovskiy gosudarstvenny universitet im. M.V. Lomonosova i

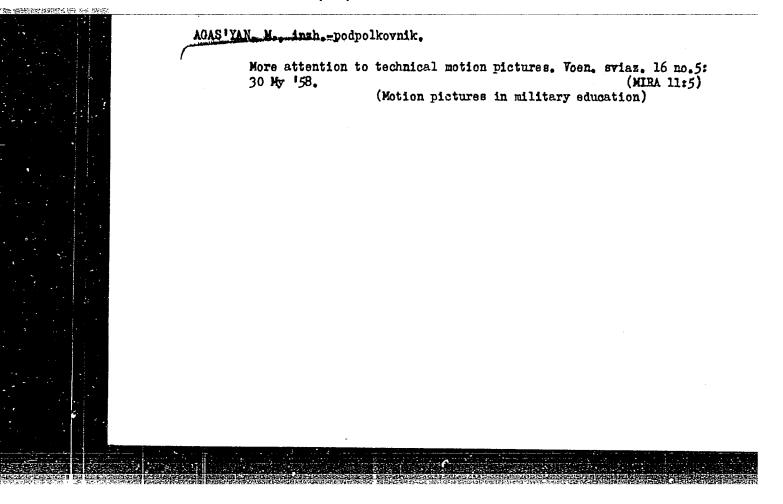
Vsesoyuznyy nauchno- issledovatel skiy institut khimicheskikh reaktivov i osobo chistykh khimicheskikh veshchestv (Moscow State University im N.V. Lomonosov and All-Union Scientific Research Institut

of Chemical Reagents and High-Purity Chemical Substances) Moscow

SUBMITTED: June 28, 1961

Card 3/3

CIA-RDP86-00513R000100510012-1" APPROVED FOR RELEASE: 06/05/2000



USOR/Phys

Magnetostriction
Invar

municam, v. n.

Nov/Dec 1947

"Temperature Dependence of the Magnetostriction of Invar Alloys,"
K. P. Belev, O. M. Agasyan, Sci Res Inst Phys, Moscow State U, 6 pp

"Izv Akad Nauk SSSR, Ser Fiz" Vol XI, No 6

Invar, elinvar, kovar, and similar alloys have anomalies of their volume and elastic properties with very complex dependence upon temperature. According to present hypotheses, the nature of these anomalies is clearly connected with the ferromagnetism of these alloys, and primarily determined by character of flow of the ferromagnetostriction phenomena. This study of magnetostriction of these alloys and especially the dependence upon temperature is of interest since it makes possible determination of the nature of the anomalous properties of Invar-type alloys.

PA 57<u>T</u>81

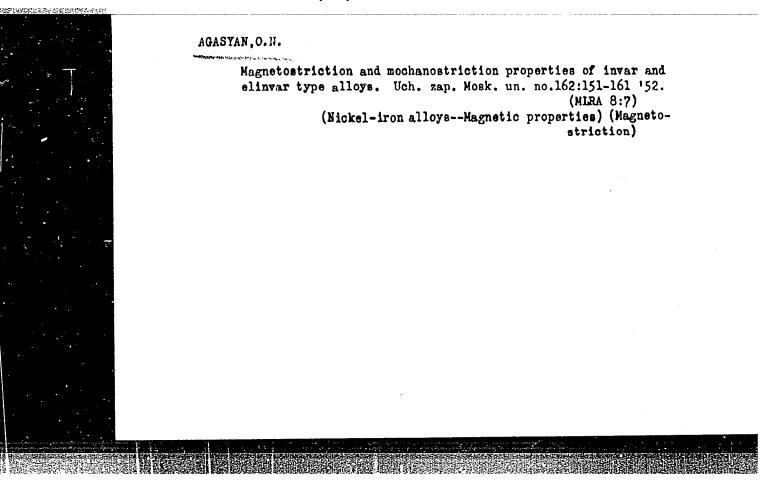
AGASYAN, O.N. ENGINEER

CAND TECH SCI

Dissertation: "Anomaly of the Modulus of Elasticity of Invar and Elinnvar Types of Alloj
28 October 49
Central Sci Res Inst of Technology and Machine Building-"TsNIITMASK."

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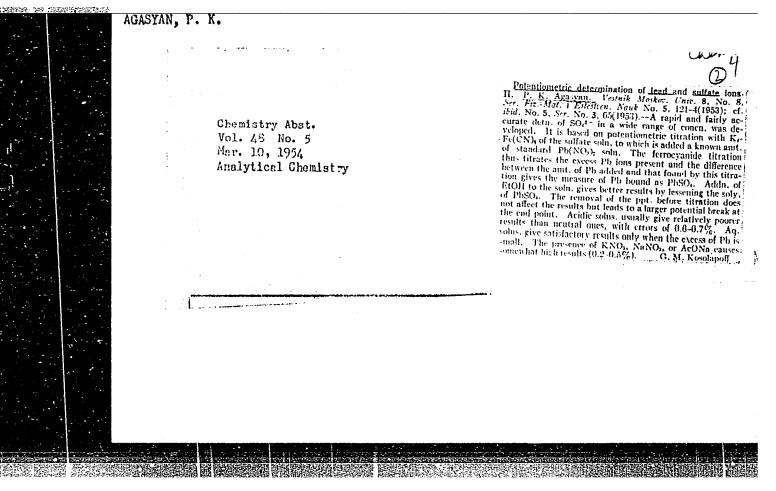
AGASYAN, O. N.	21 miles	217T45 dependence of elasticity modulus and develops formula expressing relation between temp coeff of elasticity modulus and coeff of thermal expansion. Submitted by Acad A. F. Ioffe 17 Aug 51.	SSR/Metals - Alloys, Properties 21 Oct 51 On the Nature of Elastic Anomalies in Alloys of Invar and Elinvar Types," K. P. Belov, O. N. Invar and Elinvar Types," K. P. Belov, O. N. Agasyan, Sci Res Inst of Phys, Moscow State U Agasyan, Sci Resident U Agasyan, Sci Res Inst of Phys, Moscow State U Agasyan, Sci Resident U Agasyan, Moscow State U Agasyan, Sci Res Inst of Phys, Moscow State U Agasyan, Sci Resident U Agasyan, Sci Resident U Agasyan, Sci Resident U Agasyan, Moscow State U Agasyan, Sci Resident U Agasyan, Moscow State U
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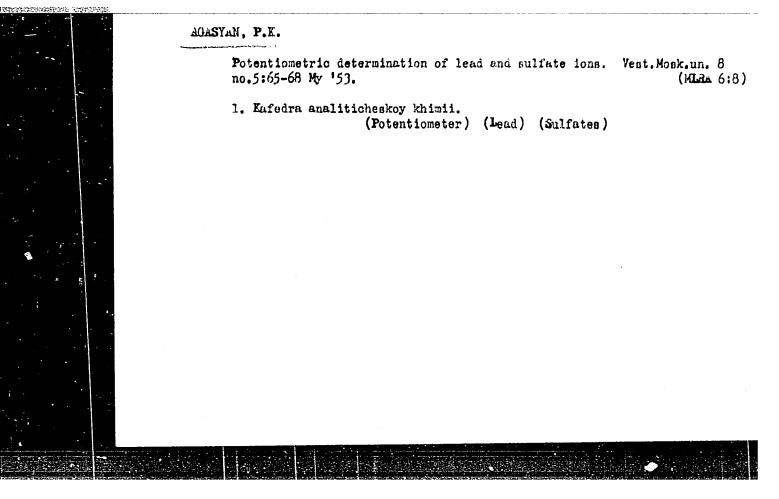


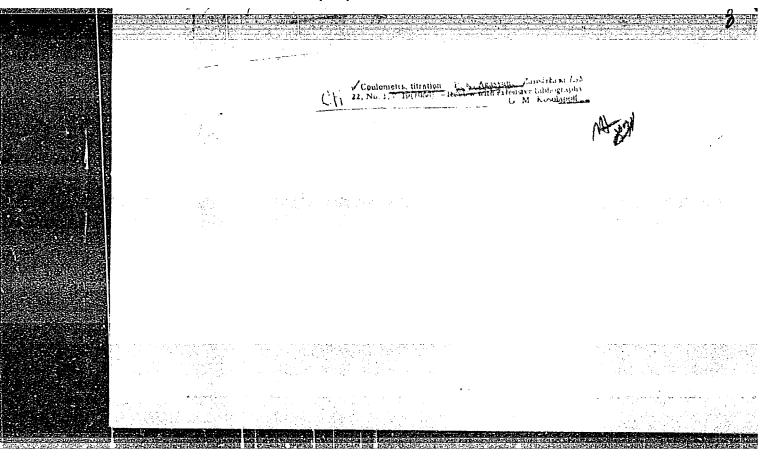
AGASYAN, P. K.

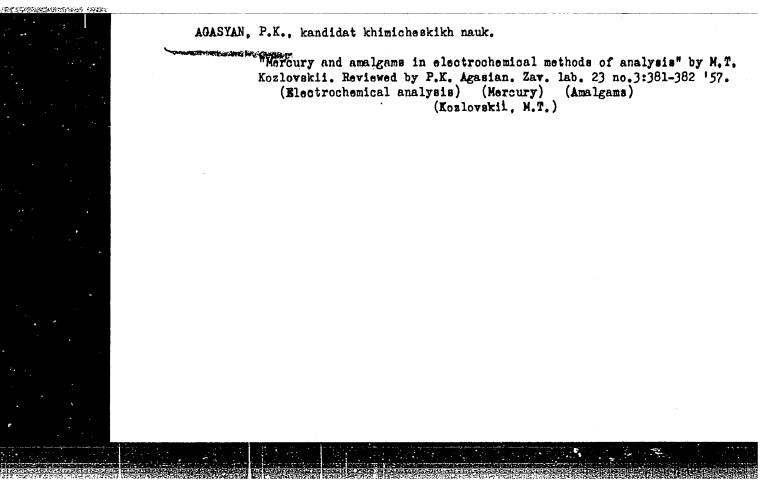
Agasyan, P.K. "Determination of lead, barium and sulfate ions by potentiometric titration," report I, Vestnik Mosk. un-ta, 1948, No. 9, p. 75-80

SO: U-2888, Letopis Zhurnal'nykh State, No. 1, 1949









AUTHOR: SOV/156-58-2-26/48 Agasyan, P. K. TITLE: Determination of Small Quantities of Indium in the Presence of Zinc (Opredeleniye malykh kolichestv indiya v prisutstvii tsinka) PERIODICAL: Nauchnyye doklady vysshey shkoly. Khimiya i khimicheskaya tekhnologiya, 1958, Nr 2, pp. 308-313 (USSR) ABSTRACT: The most effective methods of determining. indium are the spectral-, polarographic- and photometrical ones. Usually, indiam is determined by weighing as In_2O_3 , In_2S_3 , hydroxiquinolate, orthophosphate, diethyl-dithiocarbamate by means of electrolysis and the like (Refs 1-9). Further methods (Refs 6, 10-20) of titration are mentioned. The author investigated the possibility of a micro-potentiometric determination of small quantities by means of ferrocyanogen by way of compensation as referred to in the title. A modified combined micro-electrode developed by Frid (Ref 21) and the micro-buret developed by Shilov were used for this purpose. Figure 1 shows that the duration of the occurrence of an equilibrium potential increases during the titration according to the decreasing indiam concentration in the solution dard 1/3

SOV/156-58-2-26/48 Determination of Small Quantities of Indium in the Presence of Zinc

After a 3 minutes agitation the potential practically attains its constant value. It results from table 1 that the deposit takes place at a ratio of In^{3+} : $[Fe(CN)_6]^{4-} = 5 : 4$; its composition corresponds to formula $KIn_5[Fe(CN)_6]_4$ (in accordance with Refs 10, 12). The titration of different micro quantities of In^{3+} also confirms the obtained formula. The author draws the following conclusions from the results obtained: 1) The optimum. conditions of micro-potentiometric titration of indium by potassium-ferrocyanogen are the following: Room-temperature, cither a 0,05 - 0,3 M, or a 2 to 4 acetic medium. 2) The composition of indium-ferrocyanogen corresponds to formula $KIn_5[Fe(Cn)_6]_4$ in the case of titration with potassium-ferrocyanogen and to In4[Fe(CN)6]3 in the case of titration with sodium-ferrocyanogen. The minimum concentration of indium in the solution in which the titration proceeds satisfactorily, amounts to~6.0005 mol/1. 4) The molar solubility of indium- and potassium-ferrecyanogen - radiometrically determined - amounts to approximately 2,0.16 %. Thus indian and zine carnot be titrated

Card 2/3

SOV/156-58-2-26/48

Determination of Small Quantities of Indium in the Presence of Zinc

separately. 5) In the presence of 200-fold and larger quantities of zinc first of all it must be separated as hydroxide or by means of an ammonia excess. 6) Smaller quantities of indium $(>6.10^{-6} \text{ M})$ can be photometrically recorded after its extrac-

tion by chloroform-hydroxyquinoline solution at pH 3,5 also in the presence of 25 000-fold zinc-quantities. There are 4 figures, 5 tables, and 24 references, 7 of which are Soviet.

ASSOCIATION:

Kafedra analiticheskoy khimii Moskovskogo gosudarstvennogo universiteta im. M. V. Lomonosova (Chair of Analytical Chemistry of the Moscow State University imeni M. V. Lomonosov)

SUBMITTED:

October 23, 1957

Card 3/3

ACASYAN, P.K.

32-2-36/60

AUTHORS:

Lomakina, L. N., Agasyan, P. K.

TITLE:

A Combination Electrode for Potentiometrical Micro-Titration (Kombinirovannyy elektrod dlya potentsiometricheskogo mikro-titrovaniya)

PERIODICAL:

Zavodskaya Laboratoriya, 1958, Vol. 24, Nr 2, pp. 219 - 220

(USSR)

ABSTRACT:

The combination microelectrode according to B. I. Frid (reference 1) was improved by some modifications, e.g. an interposed layer of filter paper or a shift of the polished section is proposed for the purpose of preventing an etching of the polished section by crystal formation, (KCl or K₂SO₄). The interspace is in this case filled up by agar-agar and a solution of KCl. By immersing the lower part of the electrode into the electrolyte an extended storage is secured. These modifications proved to be necessary in the case of titrating small amounts according to different potentionetric methods.

Card 1/2

There are 1 figure, and 1 reference, which is Slavic.

32-2-36/60

A Combination Electrode for Potentiometrical Micro-Titration

ASSOCIATION: Moscow State University imeni M. V. Lomonosov

(Moskovskiy gosudarstvennyy universitet in. M. V. Lomonosova)

AVAILABLE: Library of Congress

1. Electrodes-Design 2. Titration-Equipment

Card 2/2

AUTHORS: Lomakina, L.N., Tarasevich, M.I., Agasyan, P.E. 32-3-6/92

TITLE: The Micropotentiometric Determination of Silver by Means of Triazoles

(Mikrojotentsiometricheskoye oprekeleniye serebra s pomoshch'yu triazolov)

PERIODICAL: Zavodskaya Laboratoriya, 1958, Vol. 24, Nr 3, pp. 270-273 (USSR)

ARSTRACT: The present paper describes a method applying benzotriazole and bromobenzotriazole for the determination of microquantities of silver; the

second-named reagent was found to be the better. For potentiometric titration a microelectrode recommended by Frid (Reference 3) in a slightly modified form was used. It was found that the potential jumps in the neutral medium are greater than in the acid medium, and that better titration results are obtained with nitric acid than with sulphuricor accided. By means of bromobenzotria cole it is possible to determine

or actic acid. By means of bromobenzotria die it is possible to determine quantities of o,ol mg/ml silver. The presence of copper, lead, nickel, cobalt, thallium and zinc does not disturb the determination in the medium of nitric acid, or in the presence of trilon B, whereas iodide-, cyanice-, and thiosulfate ions exercise a distribing effect. In weakly amioniacal solutions silver can be determined also in the presence of chloring ions.

There are several tables showing results obtained be investigation and some titration curves. There are 2 ligures, 4 tables, and 5 references,

Card 1/2 L of which are Slavic.

The Micropotentiemetric Determination of Silver by Means of Triazoles 32-3-6/52

ASSOCIATION: Moscow State University imeni M.V. Lonchosov (Moskovskiy gosudarstvennyy

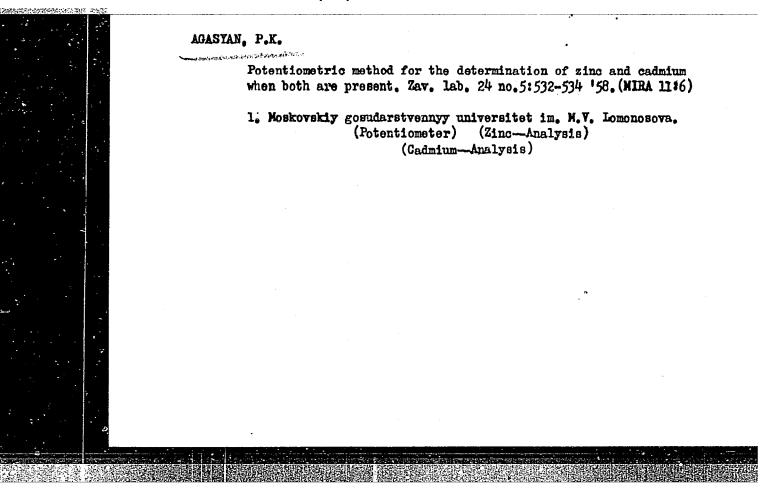
universitet im. M.V.Lomonosova)

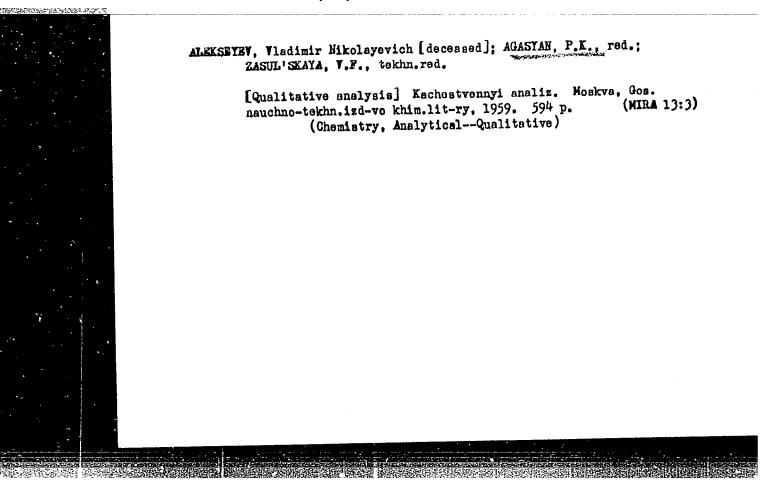
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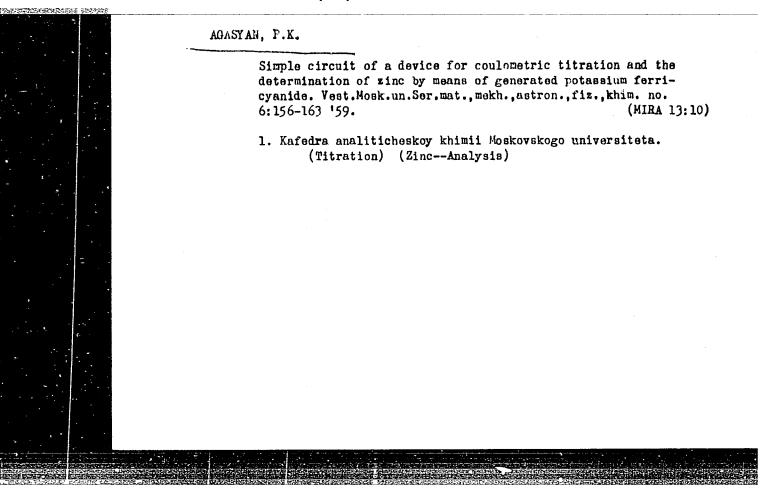
1. Silver-Micropotentiometric determination 2. Benzotriazole-

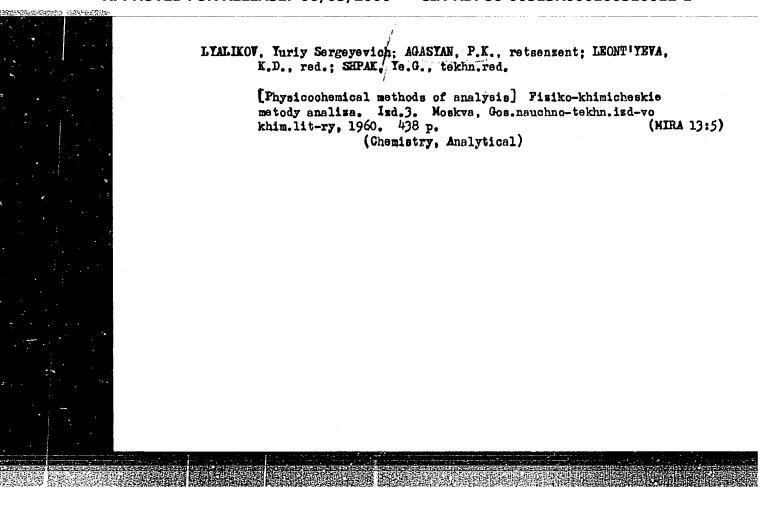
Applications 3. Bromobenzotriazele-Applications

Card 1/2



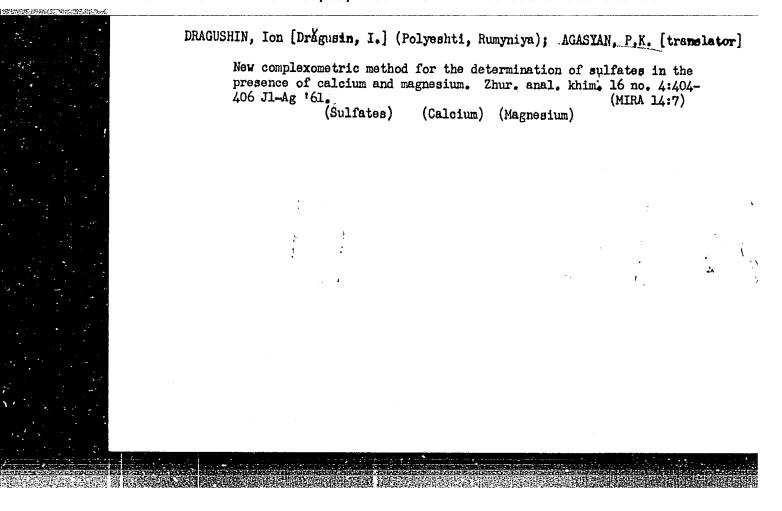






KHAKIMOVA, V.K., AGASYAN, P.K. Electrometric methods for determining tellurium (IV). Uzb. khim. zhur. no.6:21-27 '60. (MIRA 14:1) 1. Moskovskiy gosudarstvennyy universitet im. M.V. Lomonosova i Institut khimii AN UzSSR. (Tellurium---Analysis)

Whakimova, V.K.; agasyan, P.K. Use of electrolytically generated chlorine for the coulometric determination of ferrous oxide, Zav.lab, 27 no.3:263-266 '61. (MIRA 14:3) 1. Moskovskiy gosudarstvennyy universitet im. M.V. Lomonosova. (Iron oxide) (Chlorine)



PRAGUSHIN, Ion [Draglsin, I.]; AGASYAN, P.K. [translator]

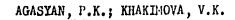
Rapid complexometric determination of barium sulfate in barites.
Zhur.anal.khim. 16 no.5:611-612 S-0 '61. (MIRA 14:9)

1. Chemical Combine Borzesti, Rumania.
(Barium sulfate) (Barite) (Complexons)

LUR'YE, Yuliy Yul'yevich; AGASYAN, P.K., red.; ZAZUL'SKAYA, V.F.,
tekhr. red.

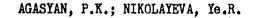
[Manual on analytical chemistry] Spravochnik po analiticheskoi
khimii. Moskva, Gos. nauchno-tekhn.izd-vo khim. lit-ry,
1962. 287 p. (MIRA 15:4)

(Chemistry, Analytical--Laboratory manuals)



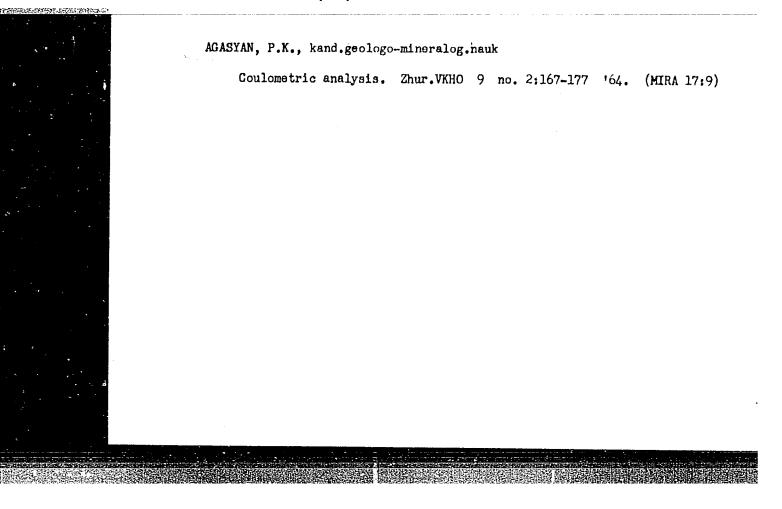
Use of certain amino acids as addends for Co (II) in its potentiometric titration by ferricyanide. Zav.lab. 28 no.10: 1184-1188 '62. (MIRA 15:10)

1. Moskovskiy gosudarstvennyy universitet imeni Lomonosova.
(Amino acids) (Cobs.lt—Analysis) (Potentiometric analysis)



Theoretical principles and interrelationship of the electrometric methods of chemical analysis (survey). Zav.lab. 29 no.7:773-781 '63. (MIRA 16:8)

(Electrochemical analysis)



AGASYAN, P.K.; NIKOLAYEVA, Ye.R.; DEMINA, L.A.

Selection of an electrometric method of determining uranium by titration with vanadate and complexon. Zav.lab. 30 no.12:1434-1438 '64. (MIRA 18:1)

1. Moskovskiy gosudarstvennyy universitet im. M.V.Lomonosova.

LEVINA, R.Ya.; GEMBITSKIY, P.A.; GUSEVA, L.P.; AGASYAN, P.K.

Cyclopropanes and cyclobutanes. Part 36: Evaluation of the reactivity of aryl cyclopropanes with the aid of Gammett equations. Zhur.ob.khim. 34 no.1:146-151 Ja '64. (MIRA 17:3)

1. Moskovskiy gosudarstvennyy universitet imeni M.V.Lomonosova.

AGASYAN, P.K.; NIKOLAYEVA. Ye.R.; RYSKULBEKOVA, R.M.

Potentiometric titration of titanium (IV) ith a solution of vanadium (II) sulfate. Zhur.anal.khim. 19 no.10:1219-1222 64.

(MIRA 17:12)

1. M.V. Lomonosov Moscow State University.

STENINA, N.I.; AGASYAN, P.K.

Coulometric titration of iridium (IV) with electrogenerated iron (II) ferrocyanide. Zhur. anal. khim. 20 no.3:351-354 '65. (MIRA 18:5)

1. Moskovskiy gosudarstvennyy universitet imeni Lomonosova.

LUR'YE, Yuliy Yul'yevich; AGASYAN, P.K., red.; ODERBERG, L.N., red.

[Handbook of analytical chemistry] Spravochnik po analiticheskoi khimii. Izd.2., perer. i dop. Moskva, Khimiia, 1965. 389 p. (MIRA 18:8)

AGASYAR, F.K.; ERYLOV, Yu.A.; Levely, A.1.

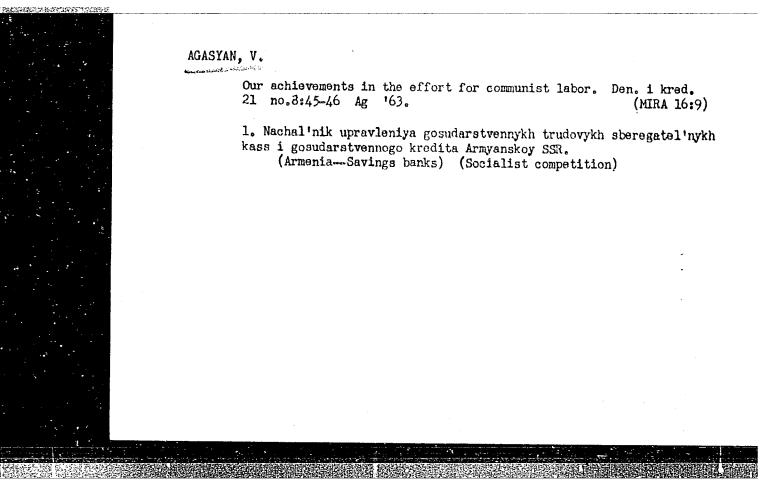
"Potentionetry" by J. Heroik, J. Tölgyessy. Poviewed by P.K.
Agasian, Tu.A. Erylov, A.1. Heroik, J. Ther. smal. khim. 20 no.6:
762 165.

(MIRA 18:7)

MASALOVICH, V.M.; NIKOLAYEVA, Ye.R.; AGASYAN, P.K.

Determination of phosphit 'n alkali metal hy comphities.
Zav. lab. 31 no. 12:1431-1434 '65. (MIRA 19:1)

1. Ural skiy nauchno-issledovatel skiy khimicheskiy institut i Moskovskiy gosmdarstvennyy universitet.



GROSUL, Ya.S., red.; ABLOV, A.V., red.; GRIHDEMG, I.F., red.;

AGAS YEVA, N.A., red.; FAYERSHTEYH, M.G., red.;

KASHUTKIN, A., red.

[From the history of science and technology; materials] Iz istorii nauki i tekhniki; materialy. Kishinev, Kartia moldoveniaske, 1963. 187 p. (MIRA 17:9)

1. Konferentsiya istorikov yestestvoznaniya i tekhniki Eoldavii. 1st, Kishinev, 1962. 2. Frezident AN Meldavskoy SSR (for Grosul). 3. Kishinevskiy gosudarstvennyy universitet (for Agas'yeva).

AGAS'YEVA, V. G.

"The polarographin behavior of metal ions in wateracetic acid and water-formic acid media." Min Higher Education USSR. Kishinev State U. Kishinev, 1955. (Dissertation for the Degree of Candidate in Chemical Sciences)

SO: Knizhnaya letopis', No. 16, 1956

·ABSTRACT:

AUTHORS: . Migal', P. K., Agas'yeva, V. G. 507/79-29-1-2/74

TITLE: The Polarographic Behavior of Metal Ions in the System Formic Acid - Water (Polyarograficheskoye povedeniye ionov

metallov v sisteme murav'inaya kislota-voda)

PERIODICAL: Zhurnal obshchey khimii, 1959, Vol 29, Nr 1, pp 3-7 (USSR)

distribution and desired and animals, 1999, 101 29, RI 1, pp 341 (055R)

The polarographic investigations in mixed solvents are of great theoretical interest as far as the various physicochemical transformations exert an influence upon the electric reduction process of ions in the media to be investigated. The application of binary fluid systems as solvents in the polarographic reduction of metal ions is expected to furnish valuable investigation material for the determination of the physico-chemical nature of the medium under investigation, especially in the case a chemical reaction is assumed to take place between the components of the system. In this connection the influence of the composition of the solvent upon the characteristic polarographic features (especially the diffusion current) of the simple metal ions was investigated in order to use the data obtained for the determination of the shares

to use the data obtained for the determination of the characteristics of the composition diagrams. A. A. Glagoleva had

SOV/79-29-1-2/74

The Polarographic Behavior of Metal Ions in the System Formic Acid - Water

> already investigated viscosity, electric conductivity, surface tension, density and fusibility of the binary system HCOOH - H2O (Refs 1-5). The analysis of the isothermal lines of the system permitted to determine the presence and the composition of two hydrates, i.e. HCOOH . H2O and HCOOH . 2H2O

> Already A. M. Zaniko and F. A. Manusova (Ref 8) and other chemists (Refs 9-11) showed that the nature of the solvent exerts an important influence upon the polarographic diffusion current. The authors investigated the polarographic character istic features of the ions Cd++, Zn++, Pb++ in the system HCOOH-H,O. The six diagrams show the dependence of the dif-

fusion current on the concentration of the metal ions. The following diagrams were obtained: the constant of the diffusion current in its dependence on the composition of the solvent. It was shown that the variation of this constant depends upon the composition of the solvent and that it is accompanied by two maxima in the concentration range corresponding to the above-mentioned hydrates. This agrees with the curves of viscosity, electric conductivity and other

Oard 2/3

The Polarographic Behavior of Metal Ions in the System Formic Acid - Water

307/79-29-1-2/74

propartias determined in the system HCOOH-H2O. There are 6

figures, 1 table, and 12 references, 8 of which are Soviet.

ASSOCIATION: Kishinevskiy gosudarstvennyy universitet (Kishinev State

University)

SUBMITTED: November 21, 1957

Card 3/3

AUTHOR:

Migal', P. K., Agas'yeva, V. G.

SOV/79-29-1-3/74

TITLE:

The Polarographic Behavior of Metal Ions in the System Acetic Acid - Water (Polyarograficheskoye povedeniye ionov

metallov v sisteme uksusnava kislota-voda)

PERIODICAL:

Zhurnal obshchey khimii, 1959, Vol. 29, Nr 1, pp 8-11 (USSR)

ABSTRACT:

In the previous paper (Ref 1) the behavior of the ions Cd++, ${\rm Zn}^{++}$, ${\rm Pb}^{++}$ in the system formic acid - water was investigated. The present paper deals with the influence of the composition of the aqueous acetic solvent upon the polarographic diffusion current of simple metal ions. The physico-chemical properties of the binary system CH2COOH-H2O were investigated by

A. A. Glagoleva, i.e. viscosity, conductivity, density, surface tension of this system. In this connection it was found that this system has an irrational character as far as on the isothermal lines of all investigated physico-chemical properties within a certain range of acid concentration the maxima and minima which point out to the formation of the hydrate complexes CH₂COOH.H₂O and CH₂COOH.2H₂O were determined.

Card 1/2

CIA-RDP86-00513R000100510012-1" APPROVED FOR RELEASE: 06/05/2000

The Polarographic Behavior of Metal Ions in the System Acetic Acid - Water SOV/79-29-1-3/74

The polarographic characteristics of the ions Cd⁺⁺, Zn⁺⁺, Ni⁺⁺ were investigated in the system CH₃COOH-H₂O. Diagrams were

plotted concerning the relation between diffusion current and composition of the solvent. It was shown that the variation of the constant of diffusion current is accompanied by clear curvatures in a section of the acid concentration corresponding to the formation of the hydrate complexes. This variation depends upon the composition of the solvent. There are 6 figures, 1 table, and 6 Soviet references.

ASSOCIATION:

Kishinevskiy gosudarstvennyy universitet (Kishinev State

University)

SUBMITTED:

October 21, 1957

Card 2/2

attalfactic, B.

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So: Honthly list of East European Accessions, (MEAL), bc, Vol. h, no. 1 dan. 1955, Uncl.

"Coffice for Projects in the Processor of the Angelian Rathres of A. 101, (1987) E.F.C. 10, Fe. 11, Forether 1984, Record, Englishman)

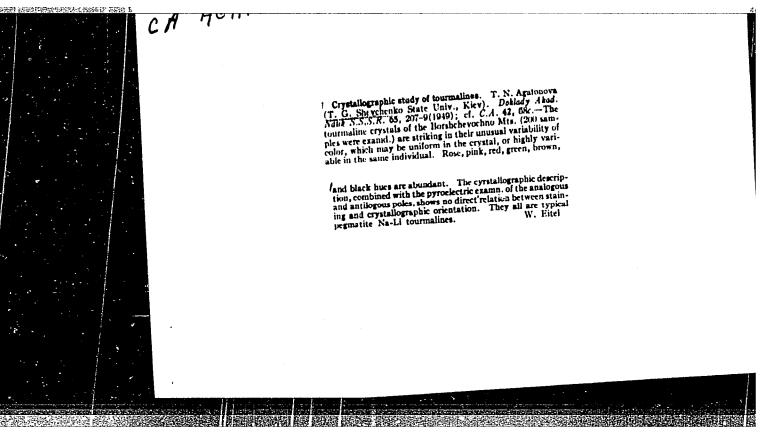
Co: Ponthly List of East European Accessions (TEAF), LC, Vol. 1, No. 3, Narch 1985, Uncl.

Technological propaganda at the Ural Heavy Machinery Factory.
Opyt rab. po tekhn. inform. i prop. no.4:29-30 '63. (MIRA 17:1)

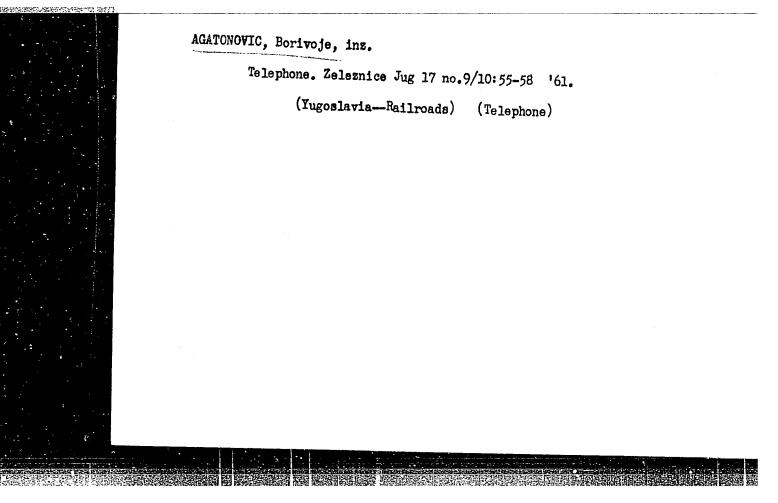
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AGATONOVIC, Evgenije, inz.

Clearness as an element of safe track crossing. Zeleznice
Jug 19 no. 2:54-58 F '63.



AGATONOVA, V.V. Late results of surgical treatment of postoperative hernias. Zdrav. Tadzh. 8 no. 2:22-24 '61. (MIRA 14:4) 1. Iz kafedry gospital'noy khirurgii (zav. - prof. N.Z. Monskev) Stalinabadskogo medinstituta imeni Abuali ibni Sino i gorodskoy klinicheskoy bel'nitsy No. 1 (glav. vrach Kh.V.Vakhidov). (HERNIA)



AGATONOVIC, E

AGATONOVIC, E. Unnecessary mediator. 1. 286

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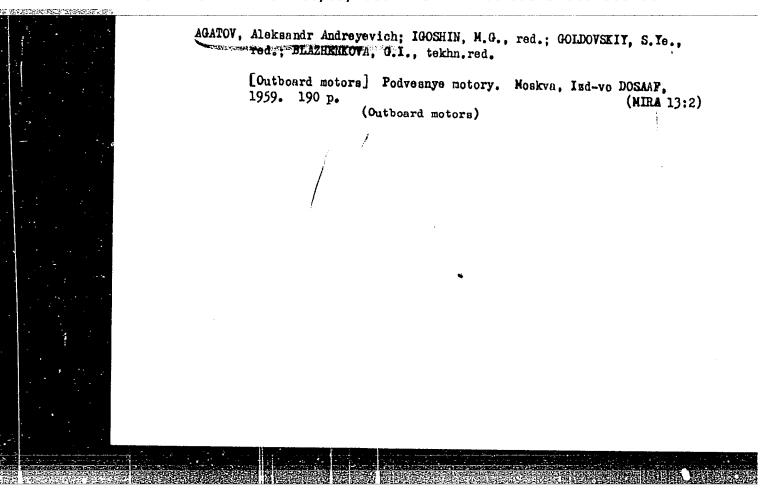
So: MONTHLY LIST OF EAST EUROFEAN ACCESSIONS, (EEAL), Vol. 4, No. 9, Sept. 1955

ACATONOVIC, E.

AGATONOVIC, E. Problems of skilled machinery and electrotechnical cadres in railroads. p. 284.

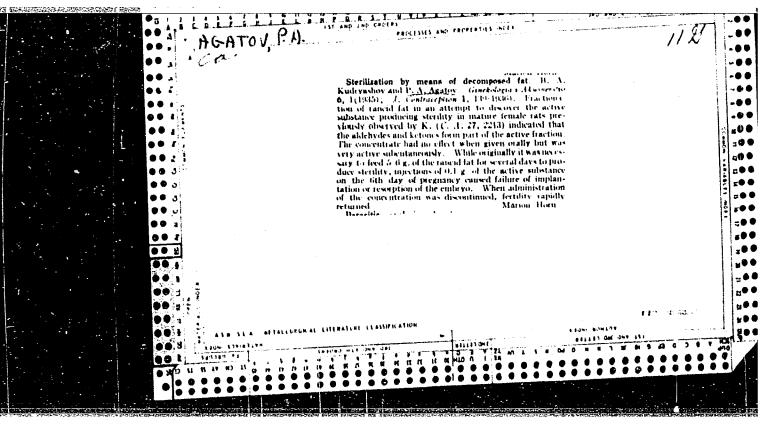
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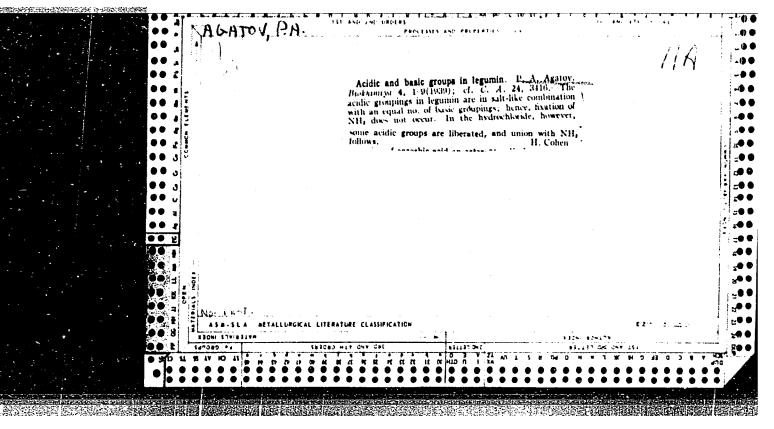
AGATOV, Aleksandr Andreyevich; KONYUSHENKO, I.A., red.; MUKHINA, Ye.S., tekhn. red.

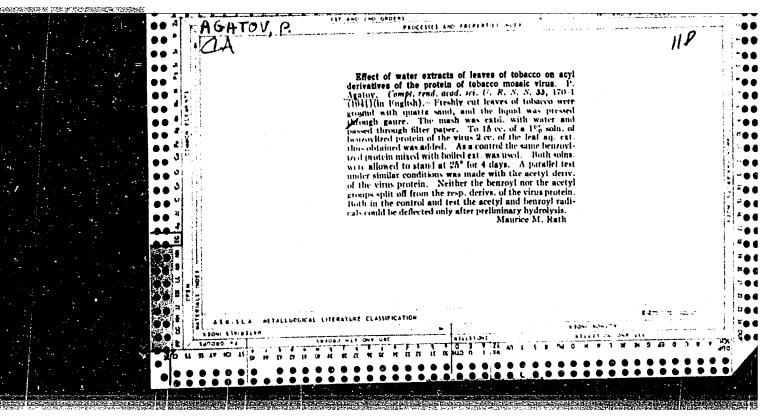
[Light fuels and their mixtures for sports motor vehicles]Legkie topliva i ikh smesi dlia sportivnykh dvigatelei. Moskva, DOSAAF, 1962. 67 p. (MIRA 16:2) (Motor fuels—Antiknock and antiknock mixtures)

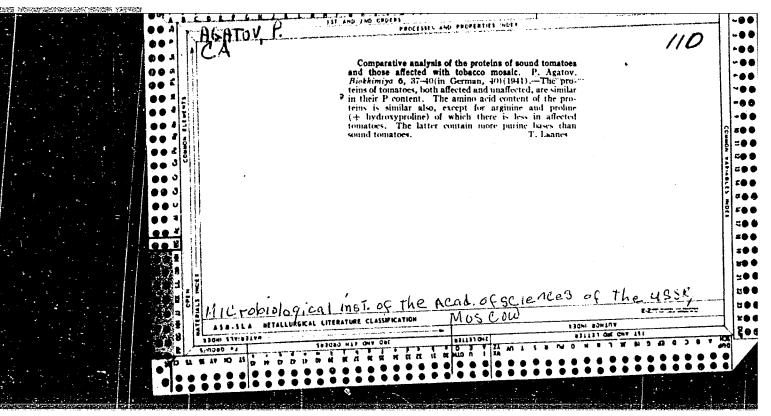


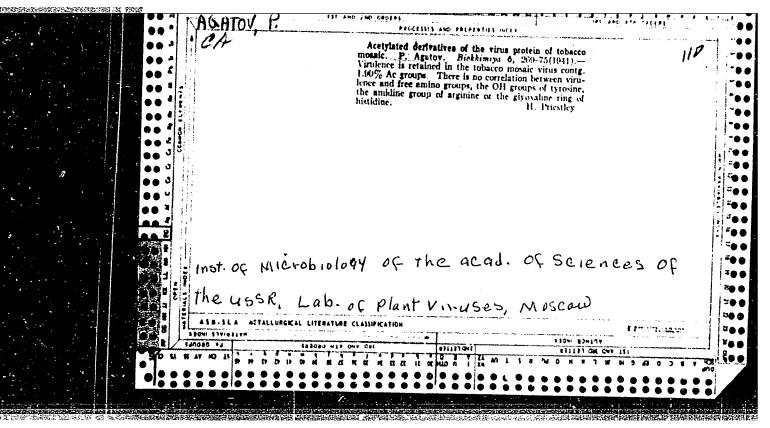
AGATOV, P.A.

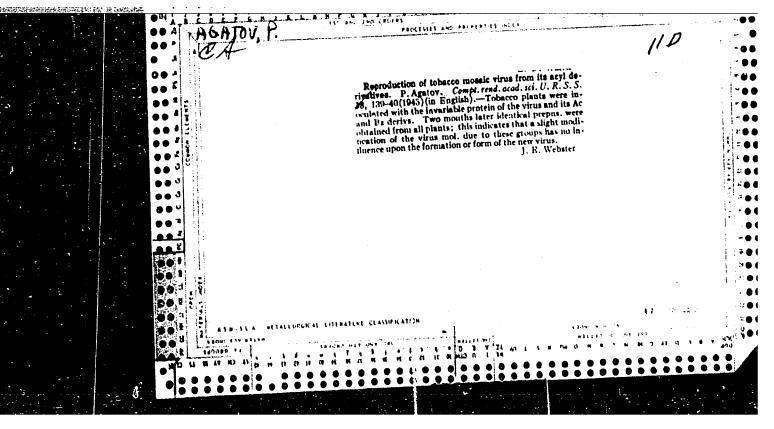
Structural chemistry of proteins. II. A. Kizel, M. P. Anamenskaya and P. A. Agatov. Trudy Lab. Isucheniyu Belka i Belkogo Obmena Or ganisme 9, 93-122(1936); Chem. Zenir. 1939, I, 141; cf. C. A. 27, 516. --As a continuation of the earlier work on the methoxylation of glycine from soybeans, the same expts. were carried out with edestin from hemp and legumes and with vicelin from peas. It was demonstrated that by treating these proteins with HCl gas in MeOH methoxylation is complete in 1 hr. In addn. to esterification, a series of side reactions took place which consisted chiefly in the splitting off of loosely bound portions of the mol. In particular, there was a decrease in the N content present as NHg, amino acids and trytophan. The combining power of the OMe and HCl groups together with a consideration of the acid groups present as detd. by titration permits conclusions to be drawn regarding the constitution of the proteins studied. It is thought that the portions of the protein mol. which are not split off are united by betaine-like linkages. Edestin which had been benzoylated by the method of Baumann was further subjected to methoxylation. It was shown that the introduction of these 2 groups took place independently of one another. W. A. Moore











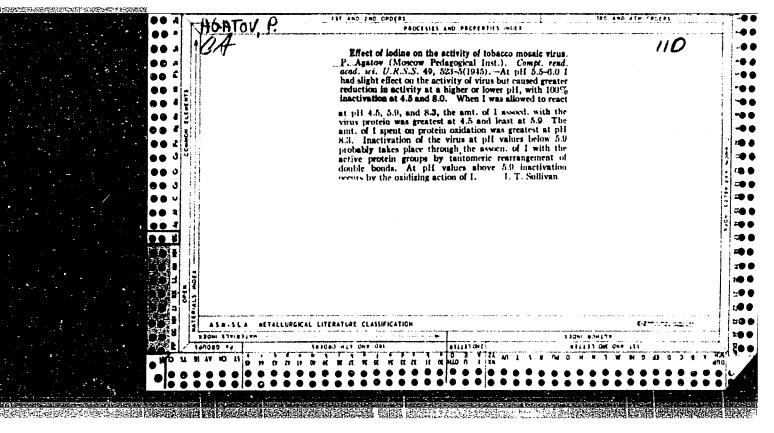
AGATOV, P. A.

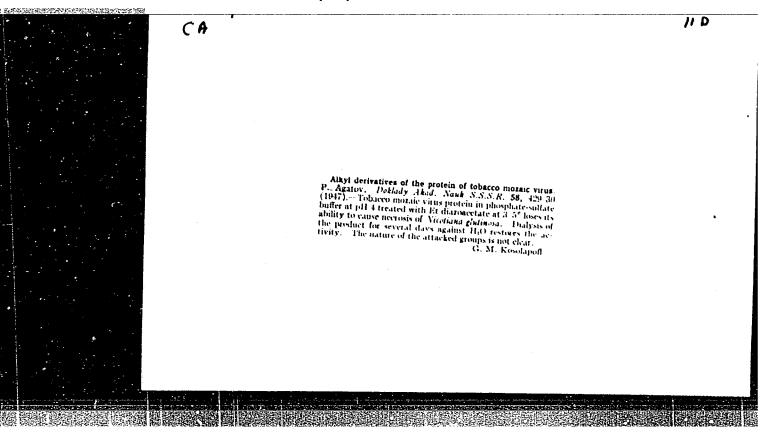
AGATOV, P. A. "Reporduction of Tobacco Mosai: Virus from its Acyl Derivatives,"

Comptes Rendus (Doklady) de l'Academie des friences de L'URSS,

vol. 38, no. 4, 1943, pp. 151-152. 511 P444

SO: SIRA, SI, 90-53, 15 Dec. 1953





USSR/Medicine - Fungi Medicine - Albumin

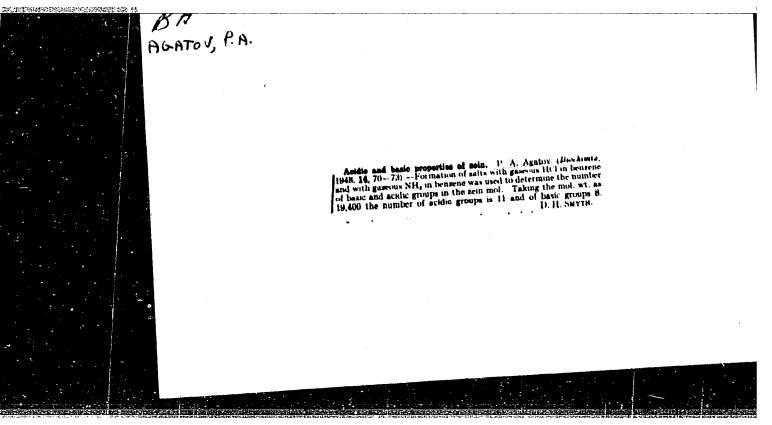
"Alkyl Producing Albumin of the Tobacco Mosaic Virus," P. Agatov, Inst Microbiol, Acad Sci USSR, 2 pp

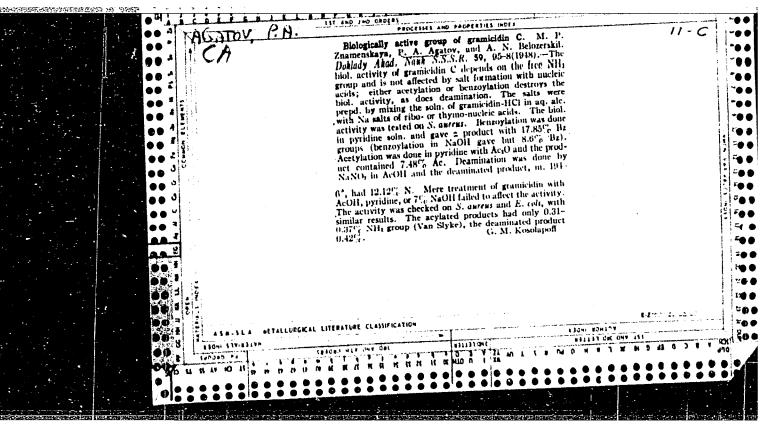
"Dok Akad Nauk BSSR, Nova Ser" Vol LVIII, No 3

For a long time scientists have had trouble determining the part that acid groups play in the infectious nature of tobacco mosaic. Agatov describes experiments and data, obtained in his work on this problem. Submitted by Academician B. L. Isachenko, 28 May 1947.

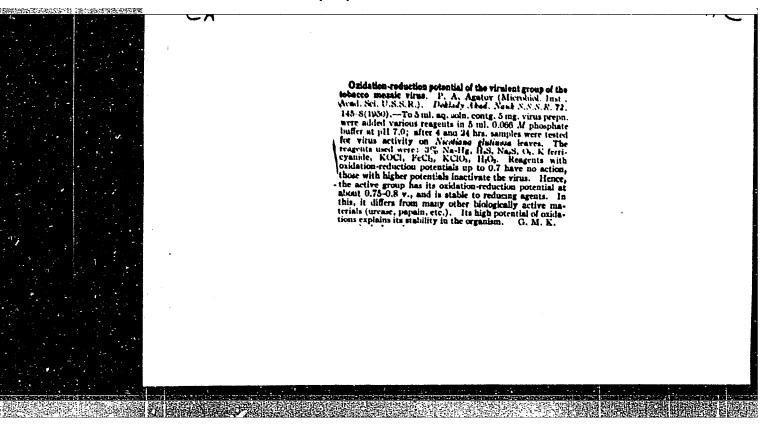
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Oct 1947





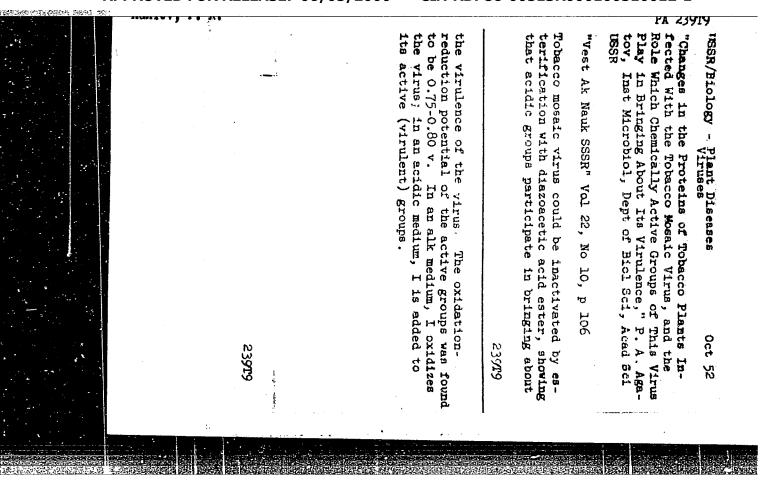
PRESENTATION OF THE TENER PROPERTY OF THE TE				
		cine - Albumin (Co sive and eight neutra base groups. Total combining groups in 1	Investigates soid and basic properties maize albumin soluble in alcohol. Salzein with gaseous HCl in benzene enabl basic groups to be studied. Similarly formation with NH3 enables a number of acid groups to be studied. Minimum zee (M 19,400) contains 11 acid groups, the	USSE/Medicine - Albumin Chemistry - Zein "Acid and Basic Properties of Zein, Inst of Microbiol Acad Sci USSE, Mo
	15T64/54	ntd) Jan/Feb 49 hlized by an equivalent number of acid- and ein molecule is 19.	properties of zein, a chol. Salt formation of zene enables all its Similarly, its salt number of its excess Minimum zein molecule groups, three of which 45/49757	Jan/Feb 49 [n, P. A. Agatov, Moscow, 4 pp
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AGATOV, P. A.

AGATOV, P. A. -- "Change of Protein Substances in the Growth of a Tomato, Affected by a Tobacco Mosaic Virus, and the Role of Chemically Active Groups of the Tobacco Mosaic Virus During Its Virulent Activity Stage." Sub 10 Jun 52, Inst of Microbiology, Acad Sci USCR (Dissertation for the Degree of Candidate in Biological Sciences).

SO: <u>Vechernaya Moskva</u> January-December 1952



AGATOV, P.A., KAZANSKAYA, T.B.,

Physiology of Actinomyces streptomycini and its relation to streptomycin production. Report No.3: Dynamics of nitrogen-containing substances in the development of Actinomyces streptomycini on a medium containing corn extract [with summary in English]. Antibiotiki, 3 no.3:28-30 My-Je 158 (MIRA 11:7)

 Institut mikrobiologii AN SSSR. (ACTINOMYCES, culture.

streptomycini, nitrogen-containing substances in cultures containing corn extract (Rus))

(NITROGEN, metabolism,

in Actinomuces streptomycini cultures containing corn extract in medium (Rus))

AGATOV, P.A.: KAZANSKAYA, T.B.

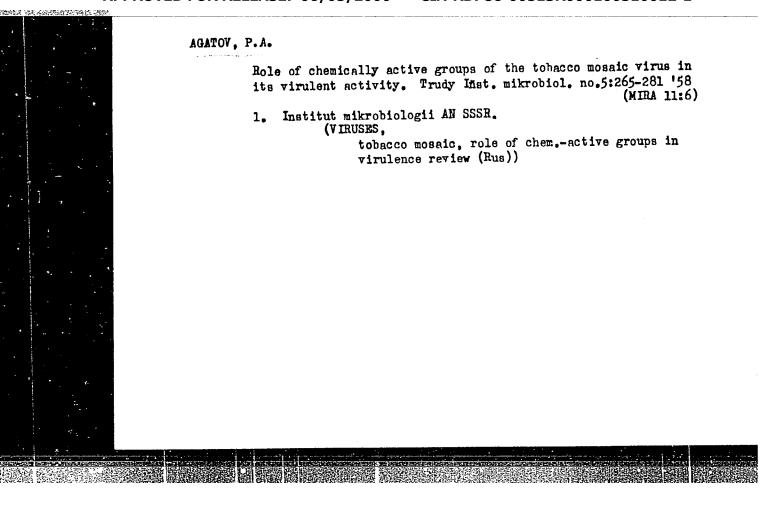
Relation of physiology to streptomycin synthesis in Actinomyces streptomycini; dynamics of nitrogen-containing substances in the development of Actinomyces streptomycini on a synthetic medium. Antibiotiki 3 no.5:31-33 S-0 158. (MIRA 12:11)

1. Institut mikrobiologii AN SSSR. (ACTINOMYCES, culture,

growth of Actinomyces streptomycini in nitrogen containing synthetic media (Rus))

(NITROGMN, eff.

on Actinomyces streptomycini growth in synthetic media (Rus))



AGATOV, P.A.; FEDOROVA, I.M.; GALANINA, L.A.

Dynamics of substances containing phosphorus in the mycelium of Actinomyces violaceus as related to its capacity for forming an antibiotic substance. Trudy Inst. mikrobiol. no. 6:245-250 '59. (MIRA 13:10)

1. Institut mikrobiologii AN SSSR. (ACTINOMYCES VIOLACEUS) (PHOSPHORUS METABOLISM)

Use of vegetable raw material in streptomycin production. Mikro-biologiia 28 no.6:858-862 N-D '59. (MIRA 13:4)

1. Institut mikrobiologii AN SSSR. (STREPTOMYCIN chem.)

17(2,12) AUTHORS: Galanina, L. A., Agatov, P. A. SOV/20-127-2-61/70 TITLE: The Effect of Some Chemical Compounds on the Formation of Streptomycin by the Strain LS-1 of Actinomyces streptomycini PERIODICAL: Doklady Akademii nauk SSSR, 1959, Vol 127, Nr 2, pp 450-452 (USSR) ABSTRACT: Complex culture media widely used in the industrial production of streptomycin like soybean flour are not suitable for the thorough investigation of the metabolism of the producer or the mechanism of antibiotic formation, because they contain complicated organic substances. On the other hand it was shown by the rechecking of the allegedly good synthetic culture media (Ref 3) that the strain mentioned in the title does not grow well on them and produces only little streptomycin. The authors repeated the experiments of Y. A. Severin (Ref 1) and I. H. Ferguson (Ref 5). The culture media of these two research workers contain soybean flour. The Ferguson method turned out to be the best. The authors believe, however, that the transfer of the seed bred on the complex culture medium to the synthetic medium to be investigated is not quite right. They used therefore Card 1/3

The Effect of Some Chemical Compounds on the SOV/20-127-2-61/70 Formation of Streptomycin by the Strain LS-1 of Actinomyces streptomycini

in both cases the same medium. Amino acids were added to the Ferguson culture medium (palculated with respect to 0.03% mitrogen centent in the culture medium) (Table 1). This shows that the streptomyoin formation is only inconsiderable without amino acids. The addition of amino acids does not only increase the growth of the mycelium, but also the formation of the antibiotic. The last process is differently stimulated by the monoamino acid (alanine on the one, and by basic amino acids (argining, histidine, lysine) on the other hand, inspite of a rather equal growth promotion. Ammonium nitrate was replaced by ammorium sulphate since it was proved that it is rather probable that the exidized form of the nitrogen does not influence at all the two mentioned processes. Since inosite (the nucleus of the streptidine part of the antibiotic) influences in general considerably the development of organisms (Ref 2) it was also added to the medium beside the amino acids. Table i shows that inssite (10 mg per 100 ml medium) does not change the mycelium growth but increases the streptomycin yield by 40-45%. Thus a culture medium was found equal to that with soybean flour. On the strength of considerations concerning the

Card 2/3

APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000100510012-1"

The Effect of Some Chemical Compounds on the SOV/20-127-2-61/70 Formation of Streptomycin by the Strain LS-1 of Actinomyces streptomycini

methyl groups of the streptobios amine part of the streptomycin molecule, and the experiments with respect to it the authors conclude that the methylation process is one of the limiting factors of the streptomycin formation in the mycelium. The methylation process is probably stimulated by an addition of vitamin B₁₂. Methicnine increases this effect since the source

of the methyl groups in the mycelium is apparently limited. There are 1 table and 5 references, 3 of which are Soviet.

ASSOCIATION: Institut mikrobiologii Akademii nauk SSSR

(Institute of Microbiology of the Academy of Sciences, USSR)

PRESENTED:

April 2, 1959, by V. N. Shaposhnikov, Academician

SUBMITTED:

March 30, 1959

Card 3/3

AGATOV, P. A., KAZ. ESKAYA, T. B., GALAHIMA, L. A., POLTAVA, I. G. (USSR)

"Participation of Certain Chemical Compounds in Streptomycin Biosynthesis."

Report presented at the 5th International Biochemistry Congress, Moscow, 10-16 August 1961

AGATOV, P.A.; NIZOVA, I.M.; QALANINA, L.A.

Distribution of nitrogenous substances in the mycelia of Act. violaceus during various intensities of antibiotic synthesis. Mikrobiologiia 30 no.5:877-830 S-0 '61. (MIRA 14:12)

1. Institut mikrobiologii AN SSSR. (ACTINOMYCES VIOLACEUS) (NITROGEN)

AGATOV, P.A.; CALANINA, L.A.; NIZOVA, I.M.

Dynamics of the phosphorus-containing substances of the mycelium of Act. streptomycini Kras. strain IS-1 at various intensity of the synthesis of streptomycin by them. Mikrobiologiia 33 no.1: 23-25 Ja-F '64. (MIRA 17:9)

1. Institut mikrobiologii AN SSSR.

AGATOV, P.A., GALANINA, I.A., NIZOVA, I.M.

Dynamics of the nitrogen compounds of the mycelium of Act. streptomycini Kras., strain LC-1, in relation to its varying rate of streptomycin synthesis. Mikrobiologiis 33 no.2: 321-324 Mr-Ap '64. (MIRA 17:12)

1. Institut mikrobiologii AN SSSR.

GALANINA, L.A.; AGATOV, P.A.

Effect of conditions helping the process of methylation on the biosynthesis of streptomycin by the IS-1 strain of Actinomyces streptomycini. Dokl. AN SSSR 166 no.1:215-218 Ja 166.

(MIRA 19:1)

1. Institut mikrobiologii AN SSSR. Submitted March 5, 1965.

ZAYTSEVA, G.N.; AGATOVA, A.I.; BELOZERSKIY, A.N.

Some data on the relationship of nitrogen fixation with respiration and oxidative phosphorylation in Azotobacter vinelandii. Biokhimiia 26 no.2:338-339 Mr-Ap '61. (MIRA 14:5)

1. The Faculty of Biology and Soil Science, State University, Moscow.

(AZOTOBACTER) (NITROGEN-FIXATION)

(OXIDATION, PHYSIOLOGICAL)

AGATOVA, A.I.; PROSKURYAKOV, N.I.

Sulfhydryl groups and disulfide bonds in wheat flour proteins. Biokhimiia 27 no.1:38-93 Ja-F '62. (MIRA 15:5)

1. Faculty of Biology and Soil Sciences, State University, Moscow.
(FLOUR) (MERCAPTO GROUP) (CHEMICAL BONDS)
(PROTEINS)

ZAYTSEVA, G.N.; KLYASHTORIN, L.B.; KHMEL', I.A.; AGATOVA, A.I.

Study of the free amino acids and amino acid composition of the protein of Azotobacter vinelandii during synchronous development. Mikrobiologiia 32 no.6:967 N-D *63 (MIRA 18:1)

1. Biologe-pochvennyy fakulitet Moskovskogo gosudarstvennogo universiteta imeni M.V. Lomonosova.